

## Groundwater Fluoride Status in Western Zone of Sanganer Tehsil of Jaipur District, Rajasthan (India)

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### Abstract:

Fluoride in groundwater causes genuine medical conditions and heaps of illnesses in numerous pieces of India. Jaipur has as of late been distinguished together of the savvy urban areas and one among numerous tehsils. Those the high fluoride content was incited to reconsider the subject and rethink the fluoride content status. Considering this, we have attempted to work out the amount of fluoride and pollution inside the groundwater parts of Sanganer Tehsil District - Jaipur, Rajasthan. Water quality in parched and semi-dry zones has been declining in the course of recent many years. Sanganer Tehsil is fighting the matter of groundwater contamination. Inside the momentum research, fluoride (F) passable fluoride content ( $> 1.5 \text{ mg L}^{-1}$ ) in refreshment chose in 25 towns in Sanganer tehsil (125 examples were gathered from the space). Water tests are basic with a pH somewhere in the range of 7.1 and 8.51. Electric conductivity (EC) goes from  $989 \text{ mhoscm}^{-1}$  to  $6105 \text{ mhos cm}^{-1}$ . Calcium hardness (ca-h) is among 45 and  $450 \text{ mg l}^{-1}$ . General hardness (T.H.) shifts 88 and  $885 \text{ mg l}^{-1}$ . Chloride goes  $80.26 \text{ mg l}^{-1}$  to  $2270 \text{ mg l}^{-1}$  and fluoride from 0.89 to  $7.12 \text{ mg l}^{-1}$ . The outcomes showed that the groundwater top-notch of a couple of districts of Sanganer tehsil was truly horrendous, undeserving for drinking, and will best be utilized after the correct cure.

**Keywords:** *Fluoride, Fluorosis, Groundwater, Pollution, Skeletal Fluorosis*

### Introduction:

Ecological contamination by fluoride can have serious ramifications for life forms, vegetation, and fauna because of water, soil, air, or a combination of each of the three mankind. The matter of high fluoride inside the climate in India and in this manner the subsequent wellbeing risks have been uncovered. The size of the worth is important to investigate factors like the wellspring of fluoride, the topographical appropriation of fluoride, and its scattering [Rajesh Kumar Yadav et al. 2012]. Soil and water assets are being dirtied on account of different natural elements like unsafe squanders, fluid and soil squanders from businesses, removal, surface hindrances, and so on By and large, water assets are

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undependable for human utilization, yet in addition for different exercises related with water system and mechanical requirements. The presence of fluoride in high fixations in food and water turned into an overall concern. Around 200 million individuals from 25 nations experience the ill effects of the staggering impacts of fluorosis [Malti Rotwar et al. 2020]. Fluoride might be a useful normal supplement found in differing focuses in air, water, and soil. At the point when devoured in ideal sums, it improves dental wellbeing, however overabundance consumption ( $>1.5 \text{ mg-1}$ ) may cause fluorosis, including dental, skeletal, and non-skeletal fluorosis optional neurological confusions [Indermitte, E. et al. 2007 and Shailaja, K. and M.E.C. Jhonson, 2007]. Reliable with safe drinking quality guidelines, the fluoride focus ought to be inside the scope of 1.00 to 1.50  $\text{mg-1}$  and past the upper level; it brings about destructive impacts on the body (WHO, 1984).

Fluorine is a most plentiful component inside the and ought to enter groundwater inside the fluoride (F-), a geochemical with fluoride-containing minerals like mycoses, hornblende, pyroxenes a consequence of the discussion by air precipitation (Jacks, G et al. 2005). High groupings of fluoride in groundwater are typically shakes in soluble groundwater - treated with low calcium or bicarbonate pop (Edmunds, W. M.. and Smedley, P. L., 2013). It additionally influences fluoride dissolvability (Edmunds, W. M., Smedley, P. L., 2013 and Ayoob, S.; Gupta, A. K 2006). For the most part, carbonate rocks go about as zinc of fluoride, and the deliberate fluoride focuses are equivalent to soil pH (Bishnoi, M et al. 2007, Chandrasekhram, D. et al. 2008, Gupta, S.C.,1991, Handa, B.K., 1988). Increased fluoride levels in groundwater because of an expanded in alkalinity in soils. Anthropogenic materials likewise assume a part in the utilization of phosphate-based composts, which regularly contain significant degrees of fluoride. (Chaney, R. L., 2012).

The danger of fluorosis is expanding quickly around the world. India faces comparable difficulties. Presently, fluoride is discovered to be high in 17 Indian locales, including Rajasthan. In Rajasthan, 22 provinces (32) presently utilize more fluoride than is permitted. Past activists (Bishnoi, M. and S. Arora 2018, Yadav, A.K., et al., 2003 ) have detailed that fluoride and fluorosis in refreshments are related with a high centralization of fluoride particles. Typically, surface water doesn't contain high fluoride; however groundwater is sullied with high fluoride content in light of the fact that the primary wellsprings of fluoride will be fluoride-containing rocks. As water goes through rocks, fluoride escapes from these stones. Subsequently, it is a wellspring of fluoride in India.

Groundwater is the start of the surface. Rocks contain up to 180  $\text{g-1}$  of sand in sandstone and up to 800  $\text{g/1}$  of stone (Sharma, D.K., 1990 ). In our investigation region, generally 40% of the populace needs refreshment from groundwater sources like hand siphons and open wells. Ingestion of antagonistic impacts of fluoride on the body liquids, particularly in the teeth and bones, just as the far and wide event of fluorosis of teeth and bones. A study was directed inside the examination region. Groundwater tests were gathered from 23 towns and towns from the investigation region.

**Study Area:**

Sanganer Tehsil is an examination region situated inside the Jaipur District of Rajasthan, India. The geographic space of Jaipur locale is 11,061.44 sq km and stretches between 26° 25' north scopes. Moreover, 27° 51' and east longitude 74° 55' and 76° 15' cover the east-focal piece of Rajasthan, isolating 13 areas into tehsils and 13 squares for managerial accommodation. Sanganer Tehsil is associated with the downtown area of Jaipur. Its elevation goes from 26° 49'N to 26° 51'N. It is between 75° 46'E and 75° 51'E. Its territory is 635.5 ch. Km, the environment of this district is semi-dry and warm with power. Temperature (15–45 °C) and precipitation 650 mm (26 inches). The number of inhabitants in Sanganer Tehsil is 573171 according to the 2011 registration.

**Methodology:**

Groundwater tests (open wells/hand siphons) are gathered at the 23 towns and states from Sanganer tehsil destinations as given in Table 1. 115 water tests (5 examples for every district) were gathered in pre-purged polyethylene. 1-liter container. The examining occurred arbitrarily in July and August 2015. Water tests were shipped off the research center for investigation utilizing standard Physico-synthetic scale strategies. The estimation is resolved as fluoride (F) with the assistance of a particular particle meter (Fluoride Analysis-Fluoride focuses in water tests) dissected with the Orion Research Ion Analyzer Model 407 A, a fluoride particle specific cathode.). Common systems are followed to recognize fluoride focuses (APHA 2005). The ionic concentrate (TISAB) used to keep up adequate ionic energy and to stay away from the mind-boggling development of good outcomes. Likewise, Physico-compound boundaries, for example, pH, EC, C-hardness, complete hardness, chloride, and alkalinity are additionally assessed to be predictable with regular techniques (APHA 2005) the comparing quality refreshment costs (APHA 2005, USPH 1985, WHO 1996, I.S.I. 1982 are given in Table 2.

**Results and Discussion**

Examination results for the different examples gathered in the investigation territory (25 towns) in Sanganer tehsil are introduced in Table 1. The work uncovered that fluoride tests of groundwater tests in 25 areas changed from 0.89 -6 7.25 mg l-1. The convergence of fluoride in 8 spots is exceptionally disturbing. The most extreme convergence of fluoride was recorded in W-10 area is 7.12 mg l-1. The passable furthest reaches of fluoride focus is 1-1.5 ppm, comparing to (WHO 1996). data have shown that hourly zones in the tehsil study territory are influenced by high fluoride focus, with more than 40% underneath as far as possible. The fluoride fixation distinction is most likely because of the distinction in the synthetic line of the stone. pH is communicated as the reach from 7.01 - 8.51. Esteem is an impression of the convergence of H+ particles inside the arrangement. The pH esteem comparable to the examination territory was found inside grades 7.01 - 8.51. The complete pH esteem

was found in the Mahal test (8.51), and the low pH (7.01) was seen from Maheshnagar (Table 1). In understanding with WHO (1996). It has been seen that pH created an immediate collaboration with fluoride filtration, demonstrating that high water levels advance fluoride spillage and hence influence groundwater (Teotia, S.P.S., et al. 1981, Wodeyar, B.K and G. Sreenivasan 1996). Fluoride fixation is additionally identified with alkalinity (Trivedi, P., 1988). Power productivity can be a value impression of the capacity of the current force catch arrangement. The U.S. suggested the admissible power age (EC) cutoff of 6105 mhoS cm<sup>-1</sup> (Table 1). Least (989 mho Scm<sup>-1</sup>) and greatest (6105 μmhoS cm<sup>-1</sup>) E.C. were accounted for from W-23 (Table 1). An immediate connection was seen between E.C. also, F as prior announced (Devi, S., et al. 2003). Water hardness isn't a choice yet maybe an adaptable and complex combination of cations and anions. The principal particle causing particles are calcium and magnesium. Calcium focus (Ca-H) goes from 45 to 450 mg l<sup>-1</sup> (Table 1). Complete hardness (TH) fluctuates from 88 to 885 mg l<sup>-1</sup> (Table 1 and Fig. 5). The WHO suggested a protected constraint of hardness, i.e., 100-500 mg l<sup>-1</sup> (Table 2). In spring water, the hardness is fundamental because of carbonates, carbonates, sulfates, and chlorides of Ca and Mg. This generally because of low liquefying fluoride. Chloride changed from 80 to 2270 mg l<sup>-1</sup> (Table 1). The higher substance of chlorides gives a pungent taste to water. The high worth of alkalinity gives an undesirable taste to water. It showed an immediate relationship with pH, F, E.C. The outcomes are in concurrence with the aftereffects of (Jain, P. et al. 2005, Hem, J.D. 1991).

The information shows that the example of groundwater gathered at different Sanganer Tehsil locales is seriously debated by undeniable degrees of fluoride and alkalinity, which is a significant danger to human wellbeing. Numerous boundaries were completely or mostly permitted. Along these lines, it was inferred that fluoride penetration principally by groundwater affected tooth and bone arrangement. In this manner, the neighborhood drink of Sanganer Tehsil can't be cooked. To deal with spring water quality, persistent preparation of Physico-compound boundaries ought to be performed. It must be utilized for cooking and drinking after past treatment. The creators unequivocally suggest that quick advances be taken for water removal, for example, The Nalgonda cycle, created by the National Environmental Engineering Research Institute. It is an interaction that includes a fast blending of water with lime (sodium or calcium carbonate), alum (aluminum sulfate), and chlorinated lime. This prompts slants and landslides. Thusly, the supernatant, which may contain just adequate measures of fluoride, is typically sifted or weakened and utilized for cooking and drinking. A few majors of nutrient C limitation on high-fat weight control plans, drink a lot of milk and eat calcium-rich vegetables like verdant vegetables. In the event that any indications of fluorosis are distinguished, stay away from the fundamental wellsprings of fluoride consumption.

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S.NO	Name of Village	pH	TDS (mg/l)	EC( $\mu$ mhos/Cm)	F (mg/l)	TH (mg/l)	Ca <sup>+</sup> H (mg/l)	Mg <sup>+</sup> H(mg/l)	cl- (mg/l)	TA (mg/l)
1	W-1	8.51 ± 0.03	1011± 12.56	1760± 2.57	3.1± 0.01	120± 1.88	55± 0.11	65± 0.12	2270± 2.31	680± 11.54
2	W-2	7.1± 0.08	1542± 23.88	2980± 1.88	7.12± 0.24	301± 0.22	211± 0.21	90± 0.49	780± 4.55	713± 9.44
3	W-3	7.5± 0.02	887±21.4 3	1354± 4.70	1.67± 1.45	136± 1.11	56± 0.01	80± 0.78	89± 2.44	656± 7.55
4	W-4	7.42 ± 0.01	754± 28.14	1280± 11.89	1.2± 0.62	126± 2.44	56± 1.34	70± 0.98	80± 2.67	543± 8.23
5	W-5	7.8± 0.11	800± 22.11	1490± 13.90	1.32± 0.98	130± 1.88	100± 2.1	30± 0.29	98± 1.34	459± 9.66
6	W-6	8.9± 0.34	760± 18.99	1260± 9.65	1.1± 0.52	185± 0.33	87± 1.92	98± 0.14	100± 4.66	760± 2.77
7	W-7	7.58 ± 0.09	650± 14.23	989± 11.40	1± 0.17	135± 1.44	60± 1.33	75± 0.59	110± 1.76	567± 8.44
8	W-8	8.11 ± 0.21	818± 22.54	1980± 4.62	1.1± 2.65	377± 2.44	188± 0.22	189± 1.28	890± 1.56	523± 10.56
9	W-9	8.1± 0.11	905± 19.92	1651± 2.44	1.13± 1.33	197± 1.99	67± 0.03	130± 0.03	170± 0.47	410± 14.01
10	W-10	8± 0.54	810± 28.88	1381± 7.88	0.89± 1.86	165± 3.55	67± 4.0	98± 1.33	98± 3.66	450± 11.32
11	W-11	8.45 ±	1368± 11.92	1989± 12.33	0.99± 2.56	499± 0.21	154± 0.12	345± 0.20	702± 0.21	656± 13.45

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		0.09								
12	W-12	8.2± 0.07	1276± 19.34	2130± 9.55	3.12± 1.03	279± 2.88	89± 1.55	190± 1.67	389± 1.45	870± 14.55
13	W-13	8.12 ± 0.18	2980± 18.89	6590± 14.01	3.45± 0.24	885± 2.76	450± 2.55	435± 2.45	1256± 4.45	1230± 21.34
14	W-14	8.4± 0.05	1389± 22.78	2890± 15.23	1.78± 0.10	645± 4.13	189± 0.11	456± 4.67	769± 1.47	1089± 18.45
15	W-15	8.9± 0.12	2021± 20.33	2900± 14.45	1.67± 0.28	300± 2.33	45± 2.88	255± 4.50	342± 0.77	1098± 19.46
16	W-16	8.21 ± 1.04	1460± 8.99	2380± 10.84	1.87± 1.23	506± 3.30	67± 1.33	439± 4.24	760± 1.45	2387± 22.76
17	W-17	8.32 ± 0.77	1112± 17.33	1870± 5.25	3± 2.87	479± 2.01	190± 1.98	289± 0.49	453± 4.22	560± 20.45
18	W-18	8.11 ± 0.56	810± 13.46	1765± 1.11	2.98± 1.90	490± 1.44	210± 0.38	280± 2.44	198± 3.22	789± 16.22
19	W-19	8± 1.89	230± 16.39	340± 10.33	1.89± 0.08	567± 3.66	189± 2.44	387± 2.28	345± 0.33	548± 11.56
20	W-20	7.5± 0.99	780± 23.78	1768± 9.51	1.34± 1.41	88± 1.66	45± 2.33	43± 1.67	80± 5.07	670± 17.42
21	W-21	8.52 ± 1.56	1320± 16.56	2367± 3.55	2.89± 1.44	286± 2.56	188± 1.30	98± 3.44	321± 4.66	650± 10.33
22	W- 22	8.11± 0.34	3210±9.3 3	6105± 7.92	1.9± 1.66	518± 1.88	178± 0.23	340± 2.70	890± 3.55	490± 12.98
23	W- 23	7.4± 0.06	1100± 10.77	1870± 8.12	1.2± 3.99	388± 2.01	198± 4.11	190± 1.42	430± 1.88	145± 10.44
24	W- 24	7.8± 1.02	810± 15.89	1280± 1.46	1.68± 2.05	120± 0.33	40± 0.01	234± 0.22	187± 1.56	512± 9.13

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25	W-25	8.34± 0.04	1090± 27.52	1985± 0.44	4.12± 1.77	407± 0.48	340± 1.22	67± 1.33	320± 0.88	430± 6.22
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Table 2: Drinking Water Parameter is given by WHO, USPH, BIS

Parameter	USPH standard	WHO standard	BIS standard
Color	Colorless	-	5
Odor	Odorless	-	Unobjectionable
Taste	Tasteless	-	Agreeable
pH	6.0-8.5	6.5-9.2	6.5-8.5
DO	4.0-6.0	-	3.0
TDS	500	500	500
TSS			100
Cl	250	500	250
SO <sub>4</sub>	250	200	200
NO <sub>3</sub>	<10	45	45
so <sub>4</sub> <sup>-</sup>	-	-	400
F	1.5	0.5	1.0
PO <sub>4</sub>	0.1		-
Ca	100	100	75
Mg	30	150	30
COD	4.0	10	-
B.O.D.	-	-	30

Table 2

<p><b>pH Analyzed for West Areas of Sanganer Tehsil</b></p>	<p><b>2 TDS Analyzed for West Areas of Sanganer Tehsil</b></p>
<p><b>EC Analyzed for West Areas of Sanganer Tehsil</b></p>	<p><b>F Analyzed for West Areas of Sanganer Tehsil</b></p>

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<p style="text-align: center;"><b>TH (mg/l)</b></p>	<p style="text-align: center;"><b>Ca+H (mg/l)</b></p>
<p style="text-align: center;"><b>TH Analyzed for West Areas of Sanganer Tehsil</b></p>	<p style="text-align: center;"><b>Ca+H Analyzed for West Areas of Sanganer Tehsil</b></p>
<p style="text-align: center;"><b>Mg+H (mg/l)</b></p>	<p style="text-align: center;"><b>cl- (mg/l)</b></p>
<p style="text-align: center;"><b>Mg+H Analyzed for West Areas of</b></p>	<p style="text-align: center;"><b>cl Analyzed for West Areas of Sanganer</b></p>

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